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Chapter 12. Capillary Action. Fluid. Surface Tension. Vaporization. the attraction of the surface of a liquid to the surface of a.... a substance that can flow and therefore take the shape of its.... a

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force that tends to pull adjacent parts of a liquid's surfac....

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in this set (23) soluble.
Capable of being

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dissolved. solution. A homogeneous mixture of 2 or more substances in a single phase. solvent. The dissolving medium in a solution.

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Gravity. Created by:
MamaUdders. Terms in
this set (23) soluble.
capable of dissolving in
a particular solvent.
solution. a
homogeneous mixture
of two or more
substances uniformly
dispersed throughout a
single phase.

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Which of the following

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is soluble in water? a.
potassium nitrate b.
silver c. benzene d.
carbon tetrachloride
_____ 12. In a solution
at equilibrium, a. no
dissolution occurs. b.

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Preparatory**

Modern Chemistry 91
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Class Date Chapter
Test B, continued 27.
Distinguish between

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ionic crystals and metallic crystals. PART V Write the answers to the following questions on the line to the left, and show your work in the space provided.

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Answer Key

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Solutions CHAPTER 12
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Answers Chapter 12
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ANSWER 1. c 2. a 3. b
2. a. alcohol b. water c.
the gels 3. The mixture
is a colloid. The
properties are
consistent with those
reported in Table 3 on
page 404 of the text.
The particle size is
small, but not too
small, and the mixture

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Solutions and
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Key**

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Particle Mass number

19. Proton 20. Neutron

21. Electron 10

CHAPTER 3 TEST

Relative charge

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ANSWER Write the

answers to the

following questions in

the space provided. 22.

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