

## Chem Final Study Guide

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### Chem Final Study Guide

the study of quantitative relationships between the amounts of reactants used and products formed by a chemical reaction, based on Law of Conservation of Mass law of Conservation of Mass In a chemical reaction, the mass of the products equals the mass of the reactants

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CHEM305-S2020 updated 05/07/2020 p. 1 of 3 Spring 2020 CHEM 305 - Study Guide for Final Exam The final exam is on Monday, May 18 from 1:00 - 2:50 pm and is worth 200 points. The final exam will be given online as a quiz in Canvas with Proctorio.

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Chemistry Study Guide for Assessment #1 DENSITY: Density = Mass / Volume Unit for density involves: unit for mass divided by unit for volume - Write some units of mass divided by units of volume: kg/m<sup>3</sup>, kg/L, g/mL 1. Know definition of density and formula Density = Mass of 1 unit volume Density = Mass / Volume 2.

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Chemistry 101 Final Exam Study Guide- Spring 2017. Acids and Bases Acid: substance that provides H<sup>+</sup> ions in water Base: substance that provides OH<sup>-</sup> ions in water. Covalent Bonds Review: ionic bonds are formed by the positive charge of the cation being attracted to the negative charge of the anion Covalent bonds: bonds formed by sharing electrons between the atoms (not ions) o 1 electron from each atom (2 atoms per bond) o Electrons available for bonding are the un-paired electrons in ...

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Chemistry exam 1 study guide. Fall 2016. Mohseni. CHEM 1120. This is a short guide of the concepts that are in the notes. It does not include how to do many of the calculations.

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weak acid Formation of a solid ...

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2factors that act as the driving force for chemical reactions o Enthalpy o Increase in disorder or randomness of the system Spontaneity Spontaneous process Processes that occur without any outside intervention Nonspontaneous Process Processes that require an outside intervention to occur Entropy S thermodynamic quantity that is measure of the randomness or ...

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