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Chapter 10: Organic Chemistry. STUDY. PLAY. Homologous Series (HS) A set of compounds whose components differ by a single repeating carbon and associated hydrocarbons. - They have similar chemical properties. - Their physical properties follow a trend (increase or decrease gradually) - In the case of alkanes, the differ by CH₂.

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Organic Chemistry Chapter 10. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. grebman. Alkenes. Terms in this set (32) Alkenes commonly undergo addition reactions and have electron rich carbon atoms (nucleophiles) that make them react with a wide variety of electrophilic reagents, T or F?

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Chapter 10: Organic chemistry - fast facts. STUDY. PLAY. Homologous series. is a series of compounds with the same general formula, where each member differs from the successive member by -CH₂-. Members of the same homologous series show a trend in their physical properties and have similar chemical properties.

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Chapter 10: Alkenes - Chemistry LibreTexts

The repeating, or monomer, units that are linked together to form nucleic acids are known as nucleotides.The deoxyribonucleic acid (DNA) of a typical mammalian cell contains about 3 × 10⁹ nucleotides. Nucleotides can be further broken down to phosphoric acid (H₃PO₄), a pentose sugar (a sugar with five carbon atoms), and a nitrogenous base (a base containing nitrogen atoms).

Chapter 10 - Nucleic Acids and Protein Synthesis - CHE 120 ...

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The replacement of only one hydrogen atom gives an alkyl halide (or haloalkane).The common names of alkyl halides consist of two parts: the name of the alkyl group plus the stem of the name of the halogen, with the ending -ide.The IUPAC system uses the name of the parent alkane with a prefix indicating the halogen substituents, preceded by number indicating the substituent's location.

10.3: Names and Properties of Alkyl Halides - Chemistry ...

10.2. Why is sulphuric acid not used during the reaction of alcohols with KI? To solve the problem, H₂SO₄ is replaced by phosphoric acid (H₃PO₄) which provides HI for the reaction and does not give I₂ as is done by H₂SO₄. 10.3. Write structures of different dihalogen derivatives of propane. 10.7 ...

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Organic Chemistry I

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