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In chemistry, thermodynamics, and many other related fields, phase transitions (or phase changes) are the physical processes of transition between the basic states of matter:

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solid, liquid, and gas, as well as plasma in rare cases.. A phase of a thermodynamic system and the states of matter have uniform physical properties. During a phase transition of a given medium, certain properties of the ...

### **Phase transition - Wikipedia**

Phase Transitions and Latent Heat. When it comes to heat

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transferring into or out of substances, changing the temperature of the sample is not all that can occur. The sample can also undergo a phase change. By "phase" we mean the solid, liquid, and gaseous states of matter. These differ primarily in how the particles involved interact with each ...

### **5.3: Heat Capacity**

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## **and Phase Transitions - Physics LibreTexts**

Pressure-induced phase transition of 1,5-diamino-1 H-tetrazole (DAT) under high pressure†. Cheng Jin† a, Ying Liu† b, Lijuan Wang a, Weijing Zhang c, Tonglai Zhang \* c and Jinlong Zhu \* ad a Center for High Pressure Science and Technology Advanced Research (HPSTAR), Beijing, 100094, China.

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### **Pressure-induced phase transition of 1,5-diamino-1 H ...**

5 Allr 2 O 11, which  
has a structural phase  
transition temperature  
near  $T_S = 210$  K.[18]  
DSC can be used to  
study structural  
changes as well due to

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the fact upon heating or cooling most phase transitions contain a substantial amount of enthalpy, however in structural transitions such as tilting of

### **Low Temperature Structural Phase Transition in the ...**

Combining these observations yields the phase diagram shown in Figure 6-5. At low temperature the phase diagram is divided into

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two phases corresponding to  $(M > 0)$  and  $(M < 0)$ . The stable phase is the one for which  $(M)$  has the same sign as  $(H)$ , but the magnetization remains nonzero even in the limit  $(H \rightarrow 0)$ .

## 6. Phase Transitions – Introduction to Statistical Mechanics

Pressure-induced structure transitions occur at pressures of



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2.3–6.6 GPa, ~8.5 GPa, and ~17.7 GPa. The phase transition at 2.3–6.6 GPa is related to the rotation of NH<sub>2</sub>, and the latter two transitions are caused by both the rotation of NH<sub>2</sub> and the distortion of the heterocycle.

#### **Pressure-induced phase transition of 1,5-diamino-1H ...**

Phase Transitions Dirac V2.4 page 8 ...A phase transition appears as a

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sharp change in the form of thermodynamic functions, as you go from one kind of behavior to another. These sharp changes are mathematical singularities. A singularity will not happen in any finite system, as in a finite liquid. The singularity can (and does) happen in an ...

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